Severe hypokalaemia with paralysis induced by small doses of liquorice

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Summary

A patient, who presented with a flaccid quadriplegia due to profound hypokalaemia, is described. Hypokalaemia and myoglobinuria were caused by the ingestion of small amounts of liquorice contained in a laxative preparation. Subsequent controlled administration of small amounts of this preparation induced marked hypokalaemia. This was associated with sodium retention and potassium loss confirming a mineralocorticoid-like action. The sodium retention was associated with suppression of plasma levels of renin and aldosterone.

Introduction

Liquorice extracts are prepared from the root of the plant *Glycyrrhiza glabra*, and contain a principal glycyrrhizic acid which has an action resembling that of mineralocorticoids. The effects of liquorice ingestion may therefore include hypertension, hypokalaemia, hypernatraemia and a fall in plasma renin and aldosterone (Molhuysen *et al.*, 1950; Louis and Conn, 1956; Salassa, Mattox and Rosevear, 1962; Gross, Dexter and Roth, 1966; Conn, Rovner and Cohen, 1968; Holmes *et al.*, 1970; Epstein *et al.*, 1977a, b; Bannister, Ginsburg and Shneerson, 1977). The dangers of injudicious use of liquorice-containing preparations are probably well known to most clinicians, although it is generally thought that substantial quantities (upwards of one g daily of glycyrrhizic acid), often for prolonged periods, are needed to cause severe metabolic disturbance.

The present paper describes a patient who

appeared unusually sensitive to a liquorice preparation, the diagnosis being confirmed by its subsequent administration under controlled circumstances.

Case history

A 70-year-old white woman had been well for most of her life apart from an above-knee left leg amputation in 1934 following a road accident. She complained of a lack of energy throughout 1973, and in November that year she was admitted to hospital for investigation. BP was 200/100 mmHg. She had a normochromic normocytic anaemia (Hb 10 g/dl), a serum potassium and sodium concentration of 1.7 mmol/l and 145 mmol/l respectively. She was given oral potassium chloride and discharged without follow-up.

In February 1974 she was re-admitted to the same hospital as an emergency, having fallen and been unable to rise. She had been unwell for the previous month with nausea, occasional vomiting and progressive muscular weakness. BP was 180/80 mmHg, serum potassium 1.1 mmol/l, sodium 140 mmol/l and urea 10 mmol/l.

Initial therapy consisted of i.v. potassium supplements, together with oral spironolactone.

She was transferred to the MRC Blood Pressure Unit 3 days later. There was a complete flaccid paralysis of all limbs, with an inability to lift the head; tendon reflexes and the cough reflex were absent and the gag reflex was very weak. BP was 182/92 mmHg, serum potassium concentration was 1.2 mmol/l, sodium 148 mmol/l, bicarbonate 33 mmol/l and urea 18 mmol/l. Serum lactic dehydrogenase was markedly elevated at 1000 u./l. Plasma

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aldosterone was low at 55 pmol/l (normal up to 500 pmol/l). Electrocardiography revealed changes typical of severe hypokalaemia with flattened T-waves and prominent U-waves. Urine was red in colour, contained no red cells and, when fractionated by gel filtration (Sephadex G75) it was shown that the red pigment was predominantly myoglobin with a trace of Hb. It was concluded that the raised serum levels of lactic dehydrogenase and the presence of myoglobin in urine resulted from necrosis of muscle.

Initial therapy consisted of i.v. potassium replacement combined with spironolactone given via nasogastric tube. Four days after admission, when serum potassium was 2.5 mmol/l, muscle biopsy suggested a toxic or metabolic myopathy with fibre vacuolation. Myoglobin disappeared from the urine within 4 days of admission. Serum potassium rose to 4.0 mmol/l and subsequent progress was uneventful.

The patient later admitted that she had for 2–3 years taken intermittently small amounts of a liquorice-containing mixture, apparently as a laxative. This medicine was found to contain equal amounts of liquorice liquid extract B.P., cascara, elixir B.P. and liquid paraffin B.P. The patient would admit to taking no more than 2 or 3 teaspoonfuls of this preparation per week. Five ml was calculated to contain 47 mg of the calcium and potassium salts of glycyrrhizic acid.

Liquorice administration under controlled circumstances

Because it appeared probable that the severe hypokalaemia was due to ingestion of liquorice, its effects were studied under control conditions.

In April 1975 the patient was placed on a fixed diet containing 134 mmol of sodium and 38 mmol of potassium daily for 25 days, during which time all urine and faeces were collected. During the first 7 days no addition therapy was given, then for 11 days the patient took her own liquorice preparation in a dose of 5 ml twice daily. This treatment was then stopped for the final period of 7 days. At no time did the patient experience symptoms. Measurements were made of serum, urinary and faecal electrolytes throughout the study (routine automated analysis). Plasma renin (Brown *et al.*, 1964) and aldosterone (Fraser, Guest and Young, 1973) concentrations were also measured before and during liquorice administration. Peripheral venous blood samples for these purposes were drawn between 8.30 and 9.30 a.m. after overnight recumbency and fasting. Exchangeable sodium and potassium were measured before and during liquorice administration by the method of Davies and Robertson (1973).

Results of special studies

Faeces were not analysed during the beginning of the run-in period, but over the last 3 days of the initial 7-day period without medication, mean output of sodium in stool and urine was 105 mmol and potassium 37.5 mmol/day. The administration of liquorice, although it did not provoke diarrhoea, enhanced the faecal output of potassium with perhaps minimal increases in faecal sodium (Fig. 1).

FIG. 1. Changes in serum sodium and potassium, and in sodium and potassium balance, induced by 11 days' medication with liquorice-containing mixture. Dietary intake of sodium and potassium indicated. Urinary output of sodium and potassium is shown by open columns; faecal output by filled and hatched columns respectively. No faecal electrolytes were available for the first 4 days of the study.

These increased faecal losses persisted throughout the 11 days in which liquorice was given. Over the same period the urinary output of sodium was markedly reduced, with a slight fall in urine potassium. The relative losses of urinary and faecal sodium and potassium were different, however, so that there was a measured net external loss of 48 mmol of potassium and a measured net retention of 228 mmol of sodium. Towards the end of the period

of liquorice administration, the patient came into sodium balance ('mineralocorticoid escape'). After stopping the liquorice preparation, urine sodium increased significantly with a concomitant fall in urinary potassium. Faecal losses reverted to the pre-liquorice pattern (Fig. 1). There was a tendency for the urinary sodium/potassium ratio to fall during the initial days of liquorice administration, but to rise slightly thereafter ('escape') and rebound to much greater than baseline value when the liquorice was stopped (Fig. 1). Total exchangeable electrolyte measurements suggested rather larger changes in electrolyte balance, exchangeable sodium increasing from 2303 to 2703 mmol and exchangeable potassium falling from 1464 to 1144 mmol. These changes probably reflect more closely the magnitude of the induced changes since daily errors in external balance data can lead to large cumulative inaccuracies over prolonged studies (Forbes and Perley, 1951; Edelman *et al.*, 1952; Miller and Wilson, 1953). Liquorice administration resulted in a marked fall in serum potassium concentrations to levels below 3 mmol/l, while serum sodium increased. Plasma concentrations of renin and aldosterone before and at the end of the period of liquorice administration were, respectively, 64 and 35 $\mu\text{u./ml}$ (normal range 20–120 $\mu\text{u./ml}$) and 110 and 55 pmol/l. BP did not change significantly during liquorice administration. Serum sodium and potassium reverted to pretreatment levels during the period following liquorice administration.

Subsequent course

In the 4 years since these studies were completed, the patient has avoided her liquorice-containing laxative and has remained well, with normal levels of serum potassium.

Discussion

There seems little doubt that the liquorice-containing medicine was responsible for this patient's severe illness with profound hypokalaemia. She has remained well at all times when avoiding the drug, while trial exposure to it under controlled circumstances provoked hypokalaemia, with potassium loss and sodium retention. Myoglobinuria in her initial severe illness was almost certainly due to hypokalaemic-induced myopathy, as has been reported previously in association with both liquorice (Gross *et al.*, 1966) and carbenoxolone (Mohammed, Chapman and Crooks, 1966; Forshaw, 1969) administration.

A remarkable aspect of the present patient was

her response to small doses of liquorice in comparison with previously reported cases.

Louis and Conn (1956) gave 7 normal subjects 2 to 5 g/day of ammonium glycyrrhizate for 3 days. They observed marked sodium retention with mild kaliuresis.

Conn *et al.* (1968) described a patient who was calculated to have taken the equivalent of about 0.5 g ammonium glycyrrhizate for 6–7 years. This man did not apparently develop hypokalaemia until first chlorothiazide and then chlorthalidone therapy was given. This same patient was given ammonium glycyrrhizate one mg/day for 2 days, followed by 2 g/day for 3 days and finally 4 g/day for 5 days. This induced sodium retention, a rise in BP, with a fall in plasma renin and in aldosterone excretion. Kaliuresis was mild, while serum potassium fell from 4.6 to a nadir of 3.5 mmol/l.

Epstein *et al.* (1977b) gave 14 normal volunteers the equivalent of 0.7–1.4 g glycyrrhizic acid daily for one to 4 weeks. This induced significant depression of plasma levels of renin, angiotensin II, aldosterone and potassium. However, the lowest plasma potassium concentration in the whole series, seen at the 1.4 g daily dose, was 3.0 mmol/l.

In the present patient, as little as 94 mg of the calcium and potassium salts of glycyrrhizic acid, given daily for 11 days, was sufficient to induce profound hypokalaemia. Despite the lack of obvious increase in urine potassium excretion over this period, the slight fall in urinary Na : K ratio with the marked rebound rise in this ratio after discontinuation of the liquorice strongly suggests a mineralocorticoid effect, although a small laxative effect appears to have occurred also. More importantly, the demonstrated rise in total exchangeable sodium with a concomitant fall in total exchangeable potassium is strongly suggestive of mineralocorticoid activity. It appeared that no more than 100–150 mg of liquorice/week, taken at home for more prolonged periods, was responsible for her severe illness. Whereas some subjects probably continue to consume liquorice surreptitiously and in large quantities even after they have been warned of the dangers, this did not seem likely in the present patient, who has avoided the drug subsequently. There seems no reason to doubt the accuracy of her estimates of the amount she previously took.

Epstein *et al.* (1977b) have shown that normal subjects vary considerably in their ability to cope with liquorice and have speculated that small doses taken regularly for prolonged periods might in susceptible persons cause severe metabolic disturbance. This seems to have been the case in the patient described here.

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